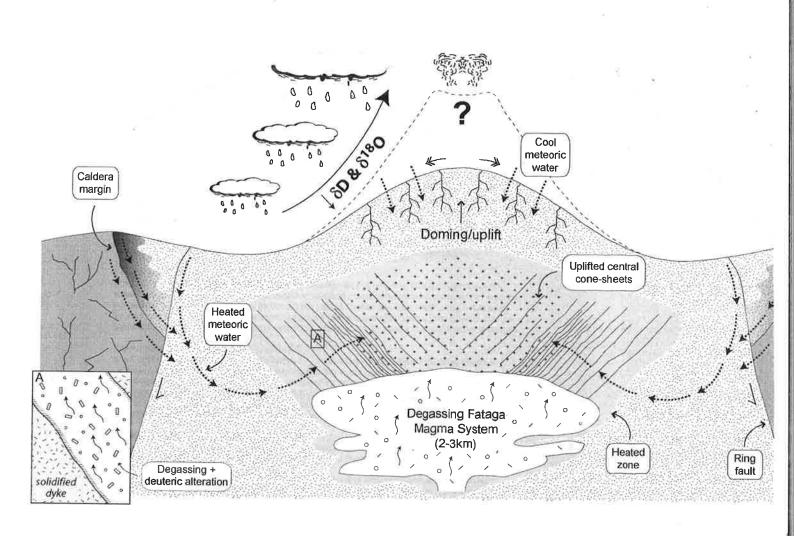
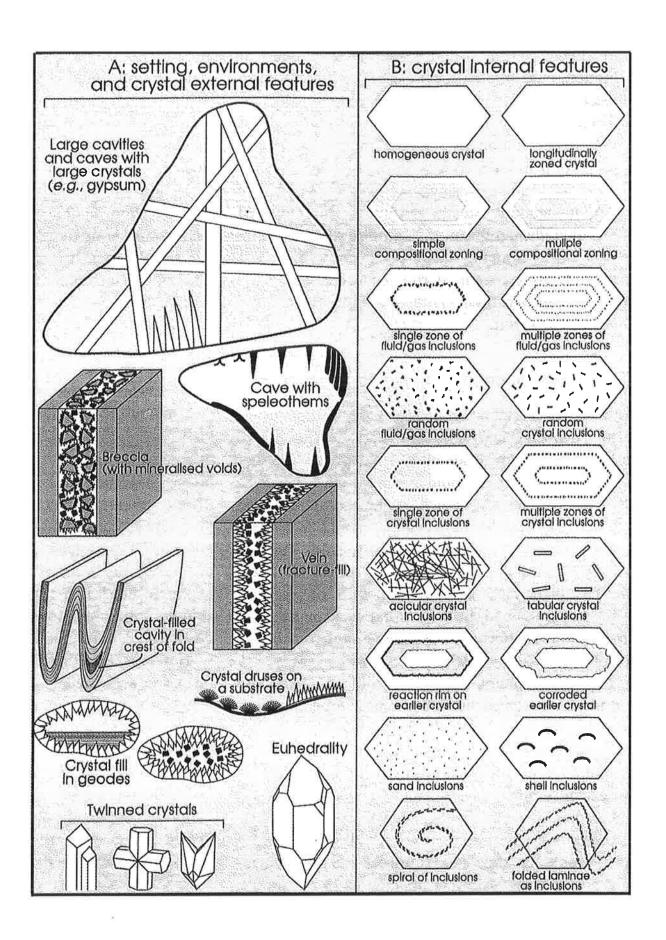
Practical Part

- Rock-forming Minerals
- Glossary of igneous textures
- Thin section petrography
 - Homeworks



Key igneous minerals

MINERAL	Hand specimen	Microscope (in PPL)	Microscope (in CPL)	Additional microscope features
	Ferro	magnesian	(Mafie)	icatures
Olivine (Mg, Fe)2 SiO4 Island silicate	Green / olive in colour	Light green to transparent	Very high interf. colours (3 rd order). Straight	No cleavage, irregular cracks, often resorbed
Clinopyroxene Ca, Mg, Si2O6 Chain-silicate	Black to brown	Brown to black to green	extinction. High interf. colours (2 nd order). Inclined extinction.	90° cleavage, sometimes zoned (8-sided)
Amphibole (Ca,Na,K)2-3 (Mg,Fe2+,Al3+)5[OH (AlSi3)O11]2 Double-chain-silicate	Black to brown	Brown to black with strong pleochroism	High interf. colours (2 nd order). Inclined extinction.	120° to 60° cleavage (diamond-shaped) (6-sided)
Biotite K(Mg,Fe2+)3[(OH)2 (Al,Fe3+)Si3O10] Sheet-silicate	Black, shiny + flaky, very soft	Brown with minor pleochroism	Very high interf. colour (3 rd order)	Birds-eye structure (finest lamellae)
	Feisic	Minerals		
Plagioclase Na[Al,Si3O8] to Ca[Al2Si2O8] Framework-silicate	White + stumpy, sometimes grey or when extremely fresh then transparent	No colour, ie. transparent. Sometimes in clusters in volcanic rock (glomerocrysts).	Pyjama stripes white – grey to black (1 st order colours)	Cleavage + zoning often present
K-feldspar (Ka,Na) [Al2SiO2O8] Framework-silicate	White, pink, often platy, Karlsbad-twins might occur	No colour, transparent	Grey to black often with perthite. If low-T then tartan twinning	Cleavage and broad twinming sometimes present
Quartz SiO2 Framework-silicate	Transparent grey, often texturally late	No colour, transparent	Grey to black. Interf. Colours of 1 st order	Undulous (patchy) extinction
Muscovite KAl2[(OH,F)2 Al,Si3O10] sheet-silicate	Silvery shiny + flacky, very soft	Grey to light brown	Intense colour of higher order	Bird-eye structure



- Notes and Comments -

OLIVINE GROUP

 $(Mg,Fe)_2[SiO_4]$

Forsterite (Fo)
Fayalite (Fa)

 Mg_2SiO_4 Fe_2SiO_4

Colour	Usually colourless. May appear pale yellow if high in Fe ²⁺		
Pleochroism	Extremely rare Fa - a + ß pale yellow and yellow		
Habit	equant, tabular, acicular, dendritic (< Mg more dendritic). anhedral plutonic subhedral/rough 6-sided phenocrysts extrusive		
Cleavage	v poor {010} {100} imperfect parting		
Relief	variable Fo moderate-high 1.635-1.670 Fa very high 1.824-1.875		
Alteration	Very susceptible (hydrothermal/ low grade meta ^m / weathering) Serpentine, chlorite, talc, carbonate, Fe oxides, iddingsite, bowlingite.		
Birefringence δ	High - =lower 3rd d (max if Fe rich)		
Interference Figures	2V very large single isogyre from isotropic section Fo ₈₅ -Fa ₁₅ - Fo ₅₀ Fa ₅₀ = 90°-75°		
Extinction	Straight		
Twinning	Rare		
Other	Zoning occassionally Mg rich may have exsolved inclusions of chromite/magnetite * Iddingsite - reddish brown RI 1.76-1.89 (smectite, chlorite. goethite/haematite) * Bowlingite - green allteration (smectite, chlorire, serpentine, talc, mica,qtz) * type depends on oxidation state of Fe		
Distinguishing features	higher d on edge of X ^I = higher Fe content Mg rich from Diopside - poorer cleavage and larger optic axial angle and higher d. Fe rich from Epidote - yellow/green pleochroism, larger optic axial angle and oblique extinction.		

PYROXENE GROUP

Magnesium- Iron Pyroxenes

Orthopyroxene – Enstatite - ferrosilite

Clinoenstatite-clinferrosilite

Pigeonite

 $(Mg,Fe)_2Si_2O_6$

 $(Mg,Fe)_2Si_2O_6$

 $(Mg,Fe^{2+},Ca)(Mg,Fe_{2+})Si_2O_6$

Calcium Pyroxene

Diopside-Hedenbergite

Augite

Ca(Mg,Fe)Si₂O₆

(Ca,Mg,Fe²⁺,Al)₂(Si,Al)₂O₆

Calcium-Sodium Pyroxene

Omphacite

Aegirine-augite

(Ca,Na)(Mg,Fe²⁺,Fe³⁺,Al)Si₂O₆ (Ca,Na)(Mg,Fe²⁺,Fe³⁺)Si₂O₆

Sodium Pyroxene

Jadeite

Kosmochlor

Aegirine

NaAlSi₂O₆

NaCrSi₂O₆

NaFe³⁺Si₂O₆

Lithium Pyroxene

Spodumene

LiAlSi₂O₆

'Normal' Px - seen in basic, calc-alkaline and in some ultrabasic and intermediate rocks. Na-Px - alkaline igneous rocks.

Exsolution lamellae: Slowly cooled Px especially Opx and augites, often contain lamellae of definite crystallographic orientation.

 $\underline{\text{E.g.}}$ Opx crystallises first at a high temperature with some Ca in the structure. It then cools and some of the Ca in the crystal is exsolved as CPx parallel to $\{100\}$ planes.

- Opx CPx lamellae is parallel to {100} planes
- Ca rich CPx exsolved Opx parallel to {100}
 · pigeonite parallel to {001}

Crystallisation trends Start A Mg rich Opx B Mg rich augite C Mg rich Ol Mg:Fe 70:30 35:65 decreasing Mg:Fe ratio

ORTHOPYROXENE Enstatite (En) – Orthoferrosilite (Fs)

Colour	Mg rich colourless	
	Fe rich pale green – pale brown	
Pleochroism	Some coloured Opx faintly pleochroic brown-yellow-green	
Habit	Early formed crystals short prismatic	
Cleavage	2 good {110} ⊥ on basal section	
Relief	Moderate- high	
Alteration	Opx → serpentine also amphibole (during which sometimes Fe	
	oxides are released)	
Birefringence δ	Low first order greys (En) – yellow/reds (Fe rich)	
Interference Figures	Large biaxial	
Extinction	Straight to edge/cleavage	
Twinning	Absent	
Others	Exsolution lamallae	
Distinguishing	OPx distinguished from CPx by parallel extinction	
features		

CLINOPYROXENE Diopside (Di) – Hedenbergite (Hed)

Colour	Di – colourless Hed – brownish green
Pleochroism	Hed – weakly pleochroic from pale green/brown (NOT
	diagnostic feaure)
Habit	Short subhedral crystals
Cleavage	{110} good. Basal intersection 87°. Partings present
Relief	Moderate- high
Alteration	Similar to Opx
	Di rarely to chlorite
Birefringence δ	Moderate – mid 2 nd order greens and yellows
Interference Figures	Moderate 2V
Extinction	Large angle - various
Twinning	Single and multiple common
Others	Exsolution lamallae
Distinguishing	Di - basic extrusives
features	Hed - acid

Pigeonite

Similar to Di and augite (2V small <30°)

 δ very low 1^{st} order greys

2 cleavages meeting at <90°

Occurring in rapidly chilled rocks. Undergoes transformation into Opx if slowly cooled.

AUGITE

Colour	Colourless to pale brown	
	Titanaugite pale purple	
Pleochroism	Very weak. Titanaugite weakly pleochroic pale green-pale	
	brown.	
Habit	Variable subhedral prismatic crystals (plutonic) → euhedral	
	(basic extrusive)	
Cleavage	Similar to diopside {110} good, partings visible	
Relief	Moderate- high	
Alteration	Similar to Diopside	
Birefringence δ	Moderate – low 2 nd order blues and greens	
Extinction	Similar to diopside	
Twinning	Similar to Diopside	
Others	Hourglass zoning especially titanaugite	
Distinguishing	Virtually indistinguishable from Di except may have smaller 2V	
features		
Occurrence	Augite mafic and ultramafic plutonic rocks	
	Diopside – metamorphic and basic volcanics	

AMPHIBOLE GROUP

Inosilicates
Orthorhombic/monoclinic

Anthophyllite - gedrite group

Ca poor (Ca & Na nearly zero) Orthorhombic and monoclinic

 $X_2Y_5Z_8O_{22}(OH,F)_2$

X = Mg, Fe,

Y = Mg, Fe, Al

Z=Si,A1

Hornblendes and tremolite-ferroactinolite group

Ca rich (Ca>Na)

Monoclinic

 $AX_{2}Y_{5}Z_{8}O_{22}(OH,F)_{2}$

A = Na

X = Ca

Y= Mg, Fe, Al

Z=Si,Al

Glaucophane-riebeckite, richterite and ockermannite- arfvedsonite group

Alkali (Na>Ca); Monoclinic

 $AX_{2}Y_{5}Z_{8}O_{22}(OH,F)_{2}$

A=Na

X= Na or (Na,Ca) Y= Mg, Fe, H Z= Si, Al

General

Colour	Green, yellow, brown (pale/strong) Mg rich - colourless or pale coloured with slight pleochroism Fe rich and alkali – strongly coloured and pleochroic
Habit	Elongated prismatic – diamond shaped cross sections
Cleavage	2 prismatic cleavages - intersection angles at 56° (acute angle)
Relief	Moderate - high
Alteration	Usually to chlorite or talc (with water)
Birefringence δ	Low to moderate – upper 1^{st} or low 2^{nd} order Fe rich high interference colours but strong body colours of alkali mask δ .
Interference figure	Large 2V except glaucophane/katophorite. Alkali not seen
Extinction	Orthorhombic – straight extinction Monoclinic - variable
Twinning	Common on {100} single or multiple
Zoning	Fairly common
Occurrence	Ca poor and Ca rich rarely seen in rocks unless metamorphosed.

HORNBLENDE Na₀₋₁Ca₂(Mg₃₋₅Al₂₋₀)(Si₆₋₇Al₂₋₁)O₂₂(OH,F)₂

Colour	Variable light brown, green, darker colours if Fe rich	
Habit	Prismatic crystals usually elongated	
Pleochroism	Variable with pale green/brown/dark green Fe rich – yellow/brown/green	
Cleavage	2 prismatic cleavages - intersection angles at 56° (acute angle)	
Relief	Moderate - high	
Alteration	See general note	
Birefringence δ	Moderate –low 2^{nd} order blues but strong body colours of Fe rich mask δ .	
Interference figure	See general	
Extinction	See general	
Twinning	See general	
Occurrence	Can appear as corona to Ol crystals in some basic rock 1 st minerals in intermediate plutonic igneous rocks Mg rich in basic, Fe rich in acidic	

Monoclinic

RIEBECKITE

Na₂(Fe²⁺Fe³⁺)Si₈O₂₂(OH)₂

Colour	Dark blue - greenish	
Habit	Large subhedral prismatic crystals or tiny crystals in groundmass	
	of alkali microgranites	
Pleochroism	Common blue/deep blue/yellow green	
Cleavage	See general	
Relief	Moderate - high	
Alteration	Common – asbestos. Found in intimate association with Na-Px	
	(aegirine) alkali granites and syenites	
Birefringence δ	Low - moderate -masked by strong body colours	
Interference figure	Occassinoally strong body colours mnake it hard to obtain	
Extinction	Length fast 6-8° {010} will give maximum angle	
Twinning	Simple or multiple on {100}	
Distinguishing feature	Dark blue and length fast	
Occurrence	Alkali igneous rocks – alkali granites associated with aegerine	

Katophorite

 $Na_2Ca(Mg,Fe)_4Fe^{3+}(Si_7Al)_{22}(OH)_2$

Dark coloured alkali intrusives associated with nepheline, aegirine and arfvedsonite

Oxyhornblende

NaCa₂(Mg,Fe,Fe³⁺,Ti,Al)₅(Si₆Al₂)O₂₂(O,OH)₂

(basaltic honblende)

Phenocrysts in intermediate volcanic or hyperbyssal

Andesites, trachytes etc

Kaersutite

 $(Na,K)Ca_2(Mg,Fe)_4Ti(Si_6Al_2)O_{22}(OH)_2$

Alkaline volcanic rock,

Phenocrysts in tracyte and other K rich extrusives, occasionally monzonites.

Eckermannite-arfvedsonite

monoclinic

 $Na_2Na(Mg,Fe^{2+})_4AlSi_8O_{22}(OH,F)_2$

Pleochroic - green/ blue green/ yellow

Moderate to high relief

Alkali plutonic rocks (Na rich) – nepheline syenites and qtz syenites association with aegirine-augite and apatite

Late stage crystalisation products`

Tektosilicates

FELDSPAR GROUP

<u>Alkali</u>

Orthoclase \rightarrow Albite

Plagioclase

Albite → anorthite

 $(K,Na)[AlSi_3O_8]$

 $Na[AlSi_3O_8] - Ca[Al_2Si_2O_8]$

GENERAL

Colour	Colourless with white or brown patches depending on whether alteration (clay minerals) has taken place	
Habit	Phenocrysts - euhedral → tabular/prismatic	
	Prismatic subhedral or anhedral	
Cleavage	2 {001} {010}, intersecting at nearly right angles on {100}	
	section. Partings occur	
Relief	Low	
	K feld <1.54	
	Plag >1.54	
Alteration	Clay minerals	
Birefringence δ	Max 1st order whites in Ca poor plag, yellows in Ca rich plag.	
Extinction	Repeated twinning – symmetrical extinction angle used to measure plag composition	
Twinning	K rich alk feld – simple Plag feld polysynthetic twinning/ repeated/ multiple twins.	
Zoning	Common in plag particularly extrusive phenocryts	
Distinguishing features	'Newcastle strip' twinning Perthites – unmixing/exsoplution intergrowths of - K in plag or plag in K.	

ALKALI

Sanidine – high Albite	Ab_{0-63}	Sanidine
	Ab_{63-90}	Anorthoclase
	Ab ₉₀₋₁₀₀	High Albite
Orthoclase- low Albite	Or_{100-85}	Orthoclase
	Or_{85-20}	Orthoclase cryptoperthites
	Or_{20-0}	Low Albite
Microcline – low Albite	Or_{100-92}	Microcline
	Or_{92-20}	Microcline cryptoperthites
	Or_{20-0}	Low albite

Colour	Colourless – opaque patches of alteration	
Habit	High temp porphyrtic - euhedral prismatic	
	Plutonic intrusives - anhedral	
Cleavage	2 {001} {010}	
Relief	Low <1.54	
Alteration	Clay minerals – limited water \rightarrow illite, excess water \rightarrow kaolin	
Birefringence δ	Max 1 st order whites greys.	
Interference figure	2V 40-65°	
Extinction	Varies depending on composition	
Twinning	Simple or microcline cross hatched	
Perthites	Na feld in K feld	
Distinguishing	Anorthoclase – 2 sets of twins – grid/hatch	
features	Extrusive only	
	Microcline - plutonic - large 2V (67°) impossible to obtain	
	Orthoclase – like qtz but IR <1.54	
	Alters easily	
	Biaxial -ve, slightly larger 2V than sanidine	
Occurrence	Alkali – acid – syenites, granites and granodiorites, felsites.	
	Orthoclase porphyries, tracytes, rhyolitres and dacites	
	Common in pegmatites and hydrothermal veins	
	Plutonic – orthoclase, microclinme and perthites	
	Extrusive - sanidines	

PLAGIOCLASE

Albite	0-10 %An	NaAlSi ₃ O ₈
Oligoclase	10-30 %An	
Andesine	30-50 %An	
Labradorite	50-70 %An	
Bytownite	70-90 %An	
Anorthite	90-100 %An	$CaAl_2Si_2O_8$

Colour	Colourless – opaque patches of alteration
Habit	Plutonic/hypabyssal -Subhedral prismatic
	Extrusive – euhedral prismatic
Cleavage	2 {001} {010}
Relief	Low >1.54
Alteration	Clay minerals – limited water → montmorillonite, excess water
	→ kaolin
Birefringence δ	Max 1 st order greys (Ab)/ yellows (An).
Interference figure	2V large and variable in sign
Extinction	Measure composition using symmetrical extinction angle of albite twins
Twinning	Multiple twinning (Newcastle strip) Na plag – narrow Ca plag – narrow and broad alternation Carlsbad - simple Pericline - repeated
Zoning	Common in extrusives – from Ca rich core → Na rich margin
Occurrence	Bytownite – ultrabasic
	Labradorite – basic
	Andesine – intermediate
	Oligoclase – acid

<u>NEPHELINE</u>

NaAlSiO₄

Feldspathoid

Colour	Colourless
Habit	Anhedral occurring in interstices between minerals. Found as
	exsolved blebs with feldspar (particularly K feld). Euhedral
	crystals are hexagonal
Cleavage	{1010} imperfect
Relief	Low
Alteration	May alter to zeolites → natrolite/ analcime or sodalite
Birefringence δ	Low - 1st order greys, small inclusions give a night sky effect
Twinning	rare
Occurrence	Characteristic primary crystallising mineral of alkali igneous
	rocks. Essential in silica deficient nepheline syenites. May be
	metasommatic in origin

SERPENTINE

 $Mg_3Si_2O_5(OH)_4$

Monoclinic Phyllosilicate

Chrysotile

fibrous

Lizardite and Antigorite tabular

Colour	Colourless to pale green				
Habit	Chrysotile fibrous parallel to x-axis				
	Lizardite and antigorite flat tabular				
Cleavage	Chrysotile - fibrous				
	Lizardite - basal				
Relief	Low				
Birefringence δ	Low to very low 1st order greys, often anomalous pale yellow				
Interference Figures	Chrysotile length slow				
Extinction	Straight on fibres, cleavage or edge				
Other	Textures can be pseudomorphous after				
	i.e olivine – mesh/ hourglass				
	pyroxene - bastite				
Distinguishing	Serpentine minerals have a lower relief and d than chlorite and				
features	fibrous amphibole				
	Chlorite often exhibits stronger d or anomalous colours				
Occurrence	Formation after alteration of ultrabasic (dunites/ pyroxenites/				
	peridotites.				

CHLORITE

Monoclinic

$(Mg, Fe^{2+}, Fe^{3+}, Mn, A)$	Al) ₁₂ [(Si, Al) ₈ O ₂₀](OH) ₁₆	Phyllosilicate
Colour	Colourless to green	\$ 1 TO 1 1 1 2
Pleochroism	Green varieties – pale green to colourless or da If Fe rich pale yellow to green	rker green
Habit	Tabular with pseudo-hexagonal shape	
Cleavage	Perfect {001} basal cleavage	
Relief	Low to moderate	
Birefringence δ	Very low some anomalous colours – deep Berl Mg rich – browns Fe rich - violet/ blue	in blue
Interference Figures	Rarely obtained	
Extinction	Straight to cleavage	
Distinguishing features	Pleochroism	
Occurrence	Formation from hydrothermal alteration of pyrand biotite	roxene, amphibole

ZIRCON Zr(SiO₄)

Colour	Colourless to pale brown			
Habit	Very squat, small square prism with terminal faces, euhedral crystals			
Cleavage	Imperfect and poor			
Relief	Extremely high			
Alteration	none			
Birefringence δ	Very high 3 rd or 4 th order			
Extinction	straight			
Twinning	rare			
Distinguishing	Tiny euhedral crystals in alk or acid plutonic rocks.			
features	Cassiterite and rutile have higher RI and δ and are more reddish			
	in thin section.			
Occurrence	Accessory mineral in all igneous rocks but essentially			
	intermediate → acid associated with biotite			

MICA GROUP

Phyllosilicates

 $X_2Y_{4-6}Z_8O_{20}(OH,F)_4$

X= K, Na (Ba, Rb, Cs, Ca)

Y= Mg Fe Al (Mn, Li, Ce, Ti, V, Zn, Co, Cu, V)

Z= Si, Al (Ti, Ge)

PHLOGOPITE

K₂(Mg,Fe)₆Si₆O₂₀(OH,F)₄

Monoclinic

Colour	Pale brown - colourless
Habit	Small tabular crystals
Pleochroism	Weak – yellow/brownish red/ green/ deeper yellow
Cleavage	Perfect {001}
Relief	Low - moderate
Birefringence δ	High 3 rd order – body colours can mask
Extinction	Usually straight
Twinning	Rare
Other	Reaction rims found in kimberlite intrusions
Occurrence	Common constituent of kimberlite
	Minor constituent of ultramafic rocks

 $\frac{\textbf{BIOTITE}}{\text{K}_{2}(\text{Mg,Fe})_{6\text{-4}}(\text{Fe}^{3\text{+}},\text{Al,Ti})_{0\text{-2}}\text{Si}_{6\text{-5}}\text{Al}_{2\text{-3}}\text{O}_{20}(\text{OH,F})_{4}}$

Colour	Brown or yellowish occasionally green
Habit	Tabular and subhedral hexagonal plates
Pleochroism	Common and strong – yellow/brown
Cleavage	Perfect {001}
Relief	Moderate
Alteration	Common in hydrothermally altered rocks → chlorite
Birefringence δ	High – very high– body colours can mask
Extinction	Nearly straight on cleavage - speckled appearance near extinction
Twinning	Rare
Occurrence	Primary crystallising in acid-intermediate plutonic rocks
	Not common in acid and intermed. hypabyssal nad extrusives

MUSCOVITE

 $K_2Al_4Si_6Al_2O_{20}(OH,F)_4$

Colour	Colourless
Habit	Thin platy crystals
Cleavage	Perfect {001}
Relief	Low to moderate (particularly in Fe is present)
Alteration	Absent
Birefringence δ	High – upper 2 nd – 3 rd order
Extinction	Straight on cleavage
Twinning	Not observable
Occurrence	Late stage component of acid igneous plutonic rocks

$\underline{\mathbf{QUARTZ}}$; SiO_2

Triagonal

Colour	Colourless
Habit	Euhedral crystal may appear as phenocrysts in acid extrsives but usually shapeless, interstital grains
Cleavage	None
Relief	Low just greater than 1.54
Alteration	None
Birefringence δ	Low – max 1 st order yellow
Extinction	Straight on prism edge
Others	May show corroded margin- reaction between quartz and liquid

Tridymite

Rare but may be found in quickly cooled igneous rocks – association with sanidine, augite, fayalitic olivine

Rhyolite, pitchstones, dacites etc

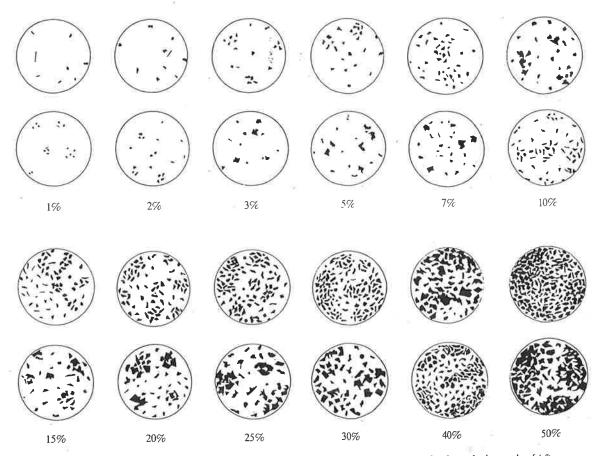
Lower RI < 1.54

Cristobalite

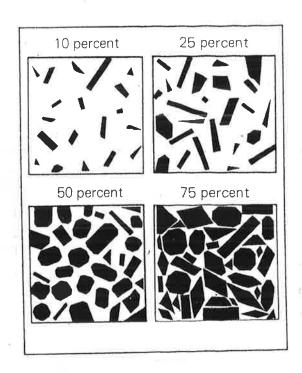
Found in cavities in volcanic rocks

OXIDES

Spinel	Composition	Colour/ opacity
Spinel	MgAl ₂ O ₄	RI=1.719. Colourless or red / brown/ blue
Magnetite	Fe ₃ O ₄	Opaque
Chromite	FeCr ₂ O ₄	Opaque – brown on edges
Rutile	TiO ₂	Reddish brown/ yellowish. Sometimes opaque - high
	-	body colour



Charts to aid the visual estimation of modal proportions of minerals in rocks [After R. D. Terry and G. V. Chilingar, American Geological Institute Data Sheet 6.]



HAND-SPECIMEN PRACTICAL

A) GRAIN SIZE:

(coarse: average > 5mm; medium; average 5mm-1 mm; fine: average <1mm)



These are averages; in general the rock is course if all the crystals can be seen with the naked eye, medium if a lens is needed to see the crystals, and fine if even with a lens the crystals are hard to see. Fine rocks are often parily or wholly glassy. Course are usually plutonic, medium are hypabystal, fine are volcanic.

B) ROCK COLOUR (M-Index)

("M;" from Greek "melanos" = black. Black is M = 100; white is M = 0).

Colour index is the proportion of dark-coloured to light-coloured minerals. The more dark minerals are present the more massic (=basic) is the rock chemistry. Exception: very fine grained or glassy rocks often appear dark independently of their mineral assemblage.

C) TEXTURE:

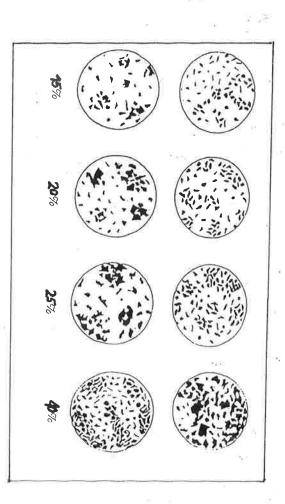
(holocrystalline/ partly crystalline/ glassy/ porphyritic/ non-porphyritic)

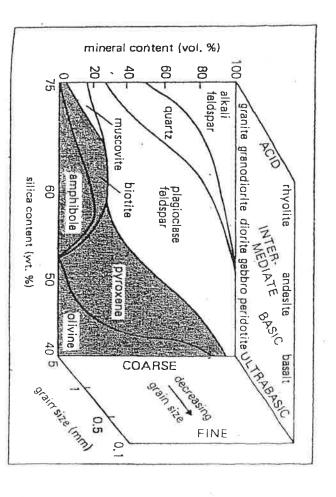
D) MINERAL CONTENT:
What minerals are present? Estimate vol%.

E) CONTENT OF LIGHT MINERALS:

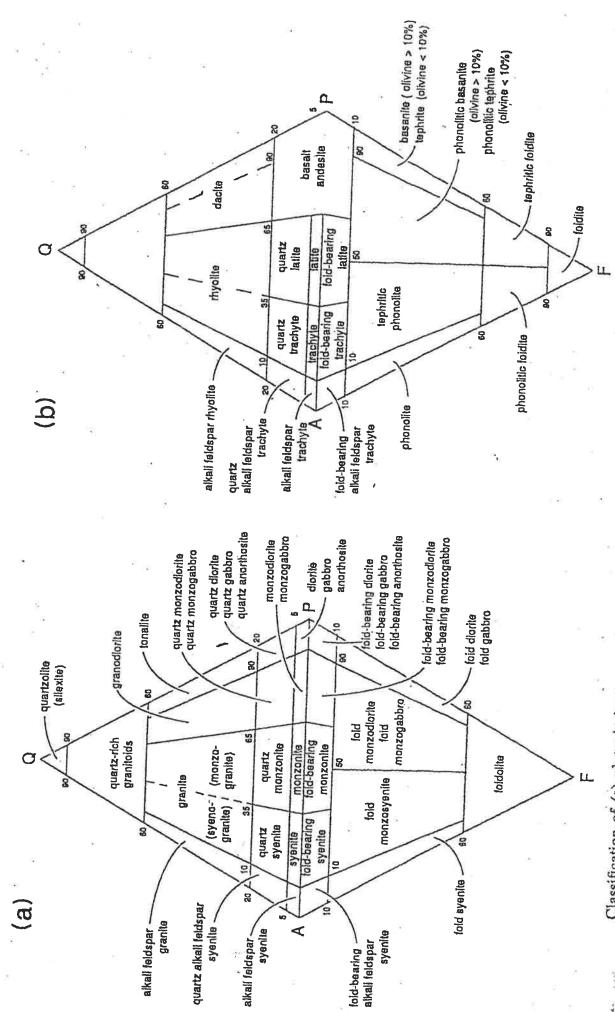
CONTENT OF DARK (FERROMAGNESIAN) MINERALS:

ROCK NAME:

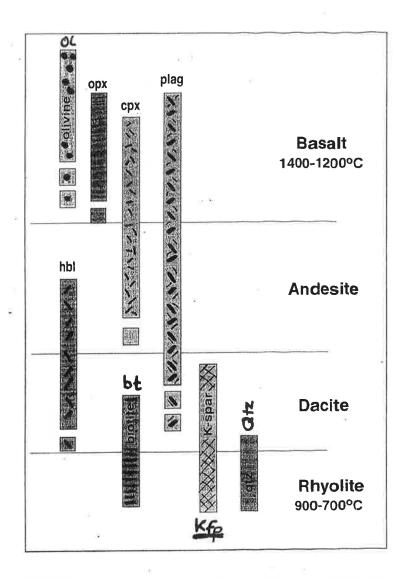




Preliminary classification scheme for igneous rocks using colour index (front face thows approximate proportions of light and dark minerals that occur at different silica contents), common silicate minerals (where recognized in the field) and grain size (decreasing with depth into the diagram).



Classification of (a) plutonic igneous rocks and (b) volcanic rocks according to the relative abundance of Q = quartz, A = alkali feldspar, P = plagioclase, and F = feldspathoid, when the mafic minerals (M) are less than 90% (Le Maitre, 1989).



Typical chemical compositions¹ for some major silicate minerals in igneous rocks (weight per cent). See Table 4.6 for distinguishing features in hand specimens of these and other minerals.

·	SiO ₂	Al ₂ O ₃	FeO + Fe ₂ O ₃	_	CaO	Na ₂ O	K ₂ O	H ₂ Q
Felsic minerals		- 5	ę.				P?	
Quartz	100	-	-				-	
Orthoclase	65	18			[4		17	-
Albite	69	19				12		-
Anorthite	43	37	-		20			
Muscovite	45	38	-				12	5
Nepheline	42	36				22		•
Mafic minerals		ST.					4	
Olivine	40		15	45		E		_
Pyroxene (augite) Amphibole	52	3	10	16	19			
(hornblende)	42	10	21	12	11	1	1	2
Biotite	40	11	16	18		-	11	4

¹Silicate minerals are made up of atomic frameworks in which different combinations of cation-forming elements are always bonded to oxygen and so it is customary to quote analyses in terms of simple oxides rather than elements.

Note also that in most mineral groups there is compositional diversity caused by the ability of atomic frameworks to hold different cations in the same site (e.g. substitution between Fe and Mg, or between Na and Ca occurs in many mineral groups).

alt / Andesite / Dacite / Trachyte Trachyte Trachyte Trachyte Microdiorite Syenite S	TEXTURE	Rhyolite / Porphyritic	Interstitial/ Microgranite intersertal +/- glass	Granite / Granular Alk. Granite	%02
Basalit / Basanite Dolerite Gabbro 45%	ROCK TYPE		1		65%
 		Andesite / Hawaiite	Microdiorite	Diorite	55%
Volcanic Hybabyssal Plutonic		Basalt / Basanite	Dolerite	Gabbro	45%
	ENVIRONMENT	Volcanic	Hybabyssal	Plutonic	

Glossary of Igneous Rock Textures

This guide is thought to help with the description of igneous rocks textures and is largely based on MacKenzie, Donaldson and Guilford "Atlas of Igneous rocks and their textures" Longman Scientific & Technical Publishers. Please bring to all practicals.

Textures may be considered to comprise four properties:

- A. Degree of crystallisation ('crystallinity'), that is the relative amounts of crystals and glass in a rock.
- B. Sizes of individual crystals ('grain size' or 'granularity')
- C. Shapes of individual crystals, and
- D. Mutual relations (ie arrangements or patterns) of crystals.

Names associated with each of these properties are listed below.

A. CRYSTALLINITY

- holocrystalline entirely crystalline.
- holohyaline either entirely glass, or mostly glass with scarce crystals (syn. 'glassy').
- hypocrystalline partly glass and partly crystals (syn. 'hemicrystalline') Notes:
- 1. Glassy can include very finely crystallised glass (ie cryptocrystalline texture, see next section).
- 2. Tiny crystals in glass are of two types: (a) those too small to have a reaction with crosspolarised light and cannot therefore be identified as a particular mineral; these form globules, rods and hair-like bodies and have the general name 'crystallites'. (b) Those which are large enough to show polarisation colours and can be identified, these have prismatic, acicular and dendritic (ie branching) shapes and are called 'microlites'.

B. GRANULARITY

General-terms:

- phanerocrystalline all crystals of the abundant minerals (>5%) can be distinguished without the microscope (ie by naked eye or with a hand lens). Phanerocrystalline includes coarsegrained and medium-grained rocks, de. Ened as having grains >5 mm and between 1-5 mm, respectively. Pegmatitic texture applies to rocks whose grain size exceeds ca. 5 cm.
- aphanitic ^{1, 2} describes a rock or the groundmass of a porphyritic rock in which no individual crystals can be distinguished by naked eye. It is effectively the same as fine-grained, defined as having grains <1 mm. Two sub-sets of aphanitic texture are: i. microcrystalline a rock or the groundmass of a porphyritic rock in which individual crystals can be identified in thin section using a rnicroscope, and ii. cryptocrystalline a rock or the groundmass of a porphyritic rock that is crystalline but the crystals can only be seen by their action on cross-polarised light and are too small to be identified as specific minerals.

A further breakdown relates to the relative grain size in a rock (see also section D):

- equigranular texture all crystals in the rock are of approximately the same size (note the, word 'approximately', ie they do not have to be exactly the same size).
- inequigranular texture there are crystals of clearly more than one size. A common example is the porphyritic texture (section D).

¹ An aphanitic rock with no phenocrysts is called **aphyric**.

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² For aphanitic rocks which are known from field evidence to be plutonic or hypabyssal some petrologists may inser the prefix 'micro'.

C. SHAPES OF CRYSTALS

Three sets of terms exist: use set I but be aware of the existence of II and III.

EXTENT OF FACE DEVELOPMENT	I	П	II
Crystal is entirely bounded by its own faces	euhedral	idiomorphic	automorphic
Crystal is partly bounded by its own faces	subhedral	hypidiomorphic	hypautomorphic
Crystal lacks any faces of its own	anhedral	allotriomorphic	xenomorphic

There also exist various terms to specify the habits or 'shapes' of crystals:

- equidimensional (syn. equant) crystal has approximately the same size measured in all directions (such as with a cube or a sphere).
- inequidimensional includes tabular, prismatic and bladed (lath) shapes.
- embayed crystal angular or round cavity or cavities penetrate the margin of the crystal; these contain glass, or devitrified glass, or secondary minerals. The term is not normally used for the junction between two primary minerals where one penetrates the other. (NB many people suppose the term to mean corroded; while embayments may form by corrosion they could also be the result of incomplete growth, ie the term has no genetic connotation)
- skeletal crystal an incomplete crystal which may lack a centre or have an intricate embayed exterior.
- dendritic crystal crystal with a more or less regular branching shape.
- pseudomorphic crystals one mineral has more or less completely replaced another but the distinctive shape of the original crystal is retained (eg serpentine occupying the prismatic shape of a now-replaced olivine).

D. MUTUAL RELATIONS OF CRYSTALS + GLASS+ ROCK FRAGMENTS

Several categories of these can be distinguished for convenience of presentation:

1. EOUIGRANULAR TEXTURES - CRYSTALLS OF ROUGELY UNIFORM SIZE

- granular :- bulk of the crystals are anhedral-subhedral (syn. allotriomorphic granular)
- subhedral granular most crystals are subhedral (syn. hypidiomorphic granular)
- euhedral granular most crystals are euhedral (syn. panidiomorphic granular)

2. INEQUIGRANULAR TEXTURES - UNEVEN GRAIN SIZE

- seriate texture crystals show a continuous range of sizes (difficult to prove without a large number of accurate measurements).
- porphyritic texture relatively large crystals (phenocrysts syn. insets) are embedded in a finer-grained groundmass or matrix³ (This is a very common igneous texture).
- glomeroporphyritic texture phenocrysts are bunched together in clots/aggregates called glomerocrysts
- polkilitic texture relatively small, roughly equant crystals of one or more minerals are scattered without common orientation in larger crystals of another mineral⁴.
- ophitic texture this is a variety of poikilitic texture in which inequidimensional crystals rather than equant ones are enclosed (eg bladed plagioclases). If both equant and elongated crystals are enclosed in the larger crystals the term polkilophitic may be used.

⁴ The term is not applied to minerals which are accessories (ie <5%) in the rock such as apatite or zircon. Neither is the term ordinarily applied to a porphyritic rock in which the phenocrysts contain inclusions of other minerals.

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³ The mineral(s) present as phenocryts may or may not be present in the matrix. A porphyritic rock can have a fine-, medium- or a coarse-grained matrix. It is named on the basis of the average grain size of the groundmass only rather than the combination of phenocrysts and groundmass; thus a rock of equal amounts of plagioclase and augite whose groundmass has an average grainsize of 0.5 mm but which contains abundant augite phenocrysts of 5-10 mm is nonetheless called a basalt. If phenocrysts have diameters of 0.5 - 0.05 mm they are called microphenocrysts and the texture is said to be microporphyritic.

- sub-ophitic texture applies when the inequidimensional crystals are only partially enclosed in the larger crystals⁵.
- interstitial texture many rocks contain wedge-shaped spaces (interstices) between randomly-arranged touching elongate crystals (eg this is common in rocks with bladed plagioclase crystals). If some of these are wholly or partially occupied by glass, or by secondary minerals that have replaced glass (eg chlorite, analcite, clays or palagonite, ie yellow-orange altered basalt glass), the term intersertal texture is used. [If elongated crystals are not touching, then glass/altered glass surrounds each grain and the term hyalopilitic texture is used (syn. vitrophyric texture). Where some crystals are touching and some are not, so that glass partially or completely encloses crystals the term hyalophitic texture may be used.] If individual interstices are completely filled with one or more or less equant grains of olivine and/or pyroxene and/or opaque the term intergranular (syn. granulitic) is used. Occasionally the interstices are empty, ie are occupied by gas, for which diktytaxitic texture is used.

3. DIRECTIVE TEXTURES

- trachytic texture a sub-parallel arrangement of bladed or tabular crystals in a fine-grained rock.
- trachytoid texture a sub-parallel arrangement of bladed or tabular crystals in a medium- or coarse-grained rock.
- banded structure at one time this term was used for trachytoid texture. If used at all these days it refers to an alternation of rock units (bands) of contrasting texture and/or relative abundance of minerals (modal mineralogy). The term has largely been displaced by 'layering'.
- eutaxitic texture a texture defined by flattened rock fragments (commonly of pumice) or glass shards in pyroclastic rocks. This is sometimes loosely referred to as banding.

4. INTERGROWTH TEXTURES

- graphic texture⁶ a regular intergrowth of quartz and alkali feldspar having the appearance of wedge-shaped Arabic writing, due to apparently isolated wedges and rods of one mineral in the other. Each intergrowth consists of one quartz and one alkali feldspar crystal, thus all the wedges and rods in the intergrowth extinguish as a single unit.
- granophyric texture a variety of graphic/micrographic texture in which the rods/wedges have a crudely radiate arrangement in the host crystal. (This texture is common in microgranites; in the UK the Geological Survey traditionally calls such rocks grartophyres.)
- spherulitic texture a radiate arrangement of very fine fibres of minerals, commonly quartz and feldspar.
- symplectite texture microscopic scale intergrowth of two minerals in which one forms sinuous, wormshaped rods in the other. Usually carries the genetic assumption that has formed by a reaction after the magma has solidified, ie it is a secondary texture formed in the solid state.
- myermekitic texture a specific combination of minerals in a symplectite intergrowth, worm-shaped rods of quartz in a plagioclase grain. Commonly located at the margin of the grain.
- exsolution (syn. unmixing) texture an intergrowth of two minerals in which one forms parallel lamellae or rods in a grain of the other (eg plagioclase lamellae in an alkali feldspar grain, as in perthite). Forms after magma has solidified, ie in solid state.

5. CAVITY TEXTURES

- vesicular texture spherieal/sub-spherieal eavities (ves~cles) in a rock. [f very abundant such that the rock is frothy in appearance, the terms scoriaceous (for basalt and andesite) and pumiceous (for dacite and rhyolite) apply.
- amygdaloidal texture vesicles completely or partially infilled with secondary minerals⁷.
- miarolitic texture miarolitic cavities are irregular-shaped holes in plutonic and hypabyssal rocks lined with euhedral quartz and feldspar.

⁵ Porphyritic, poikilitic and ophitic textures are sometimes collectively categorised as hiatal textures indicating that there is a non-continuous, ie broken, range of grainsizes in the rock.

⁶ If the rock is fine-grained, the term micrographic texture (syn. micropegmatitic texture) is used.

⁷ Sometimes the contents of an amygdale are arranged in concentric bands of two or more minerals causing the amygdale to look like a small eye in section. The term ocellus describes this pattern and a rock with many such amygdales (called ocelli) is described as ocellar.

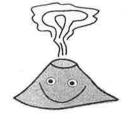
6. OVERGROWTH TEXTURE

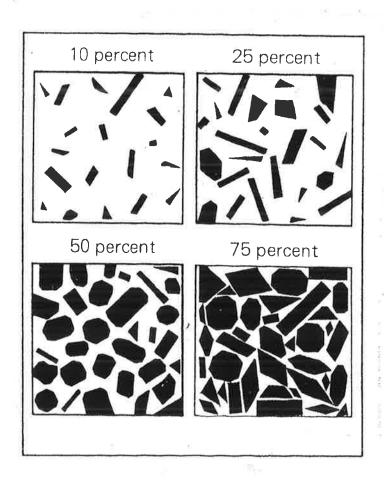
- corona texture as seen in individual crystals, this consists of concentric band(s) of one or more mineral(s) more or less surrounding another (eg an olivine core surrounded by pyroxene, possibly in turn surrounded by hornblende). The term does not apply when the surrounding mineral is the same mineral, differing only solid solution composition.
- crystal zoning variety of corona texture in which the successive stages of growth are picked out by gradual or abrupt changes in the solid solution composition of the crystal, eg a plagioclase crystal with a core of Ango surrounded by a rim or mantle of An75. Adjectives are applied to describe the zoning as continuous [steady change], or discontinuous [abrupt change(s)], normal [progressing outwards from high temperature type to low-temperature type], reverse [the opposite], and sector, or hour-glass [a complex style in which different portions of the crystal have different compositions; these portions are arranges in such a manner as to create a pattern resembling that of an hour-glass]. Zoning has also been used to describe the situation in which successive stages of growth are picked out by microscopic/sub-microscopic inclusions arranged in bands parallel to the faces of the crystal.

Addenda

- With the exception of eutaxitic texture, the above terms cover non-fragmental igneous rocks. Remember that pyroclastic rocks have a nomenclature based on average fragment size.
- Angular-rounded fragments are sometimes found in crystalline igneous rocks and several words exist for these, including **xenolith** (literally 'foreign rock'), **autolith** (syn. **cognate xenolith**) [fragments of rock genetically related to the host rock, possibly as an early-formed rind on a chamber, the rind subsequently being fragmented when some of the residual magma exited the chamber], **nodule**, and **enclave**. The latter two terms have no genetic connotations. Foreign crystals are called **xenocrysts**.
- A special, genetic vocabulary exists for the textures of rocks believed to have formed by the concentration of crystals from magmas, so-called cumulate rocks (see for example Cox et al. Interpretation of Igneous Rocks).

Have fun!

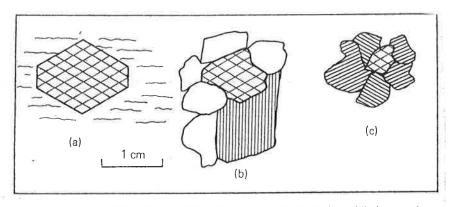




Thin Section Petrography

SEOUENTIAL CRYSTALLISATION is the distinguishing feature of igneous rocks and is seen in both Volcanic (porphyritic) and Plutonic (granular) rocks. It leads to a variety of crystal shapes and types of junction between crystals.

- In an igneous rock the crystals grow from a liquid which initially imposes no constraint on their growth, so they develop their own shapes with well developed crystal faces; such crystals are called **euhedral** crystals.
- Eventually, as space becomes used up, the crystals will meet one another and the growth of adjoining faces will be impeded. They will then interlock as growth of free surfaces continues, and will have some original faces and some inter-grown ones. Such crystals are called **subhedral** crystals.
- Later minerals will have to fit into the remaining spaces with whatever shape is available, thus being permitted very few of their own crystal faces. Such crystals are called **anhedral** crystals.



Sketches of (a) euhedral, (b) subhedral and (c) anhedral amphibole crystals

The intergrowth textures are characteristic of igneous rocks. Since liquids don't transmit directional stress, the crystals can grow randomly to produce an isotropic fabric. This is different to the anisotropic, oriented, fabric typical of metamorphic rocks which grow in a solid rock under tectonic stress.

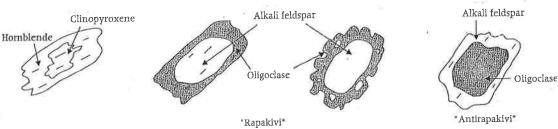
There are three special circumstances in which igneous minerals can orient themselves:

- Flow of liquid during crystallisation elongated or tabular crystals can align themselves to the flow (like logs in a river). This is most-often seen in dykes and sills in which feldspars align themselves to the magma flow through the narrow fracture.
- Settling of platy minerals to the bottom of a magma chamber
- Growth of elongate minerals perpendicular to a boundary surface producing a comb-like fabric (like the teeth of a comb). This is seen mainly in hydrothermal veins.
- If an igneous rock displays a directional fabric which is not caused by one of these methods, it has probably been involved in deformational metamorphism during growth a situation seen most often in granitoid rocks which cool slowly during orogenesis.

Some aspects of texctural development in igneous rocks

Discontinuous Series (ferromagnesian phases)

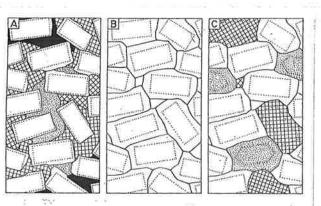
Euhedral minerals continue to grow over the crystallisation temperature range, but may begin to dissolve as temperature falls below this range - this can produce corroded crystals, but since the mineral is attempting to convert to the next in the series, it may develop a mantle of this new mineral (seen as a rim in thin section) which effectively cuts it off from the liquid and stops the conversion.



Crystal settling

In basalt magmas much of the olivine is usually crystallised early and settles to the bottom of the magma chamber, so most gabbros, shallow intrusions and lavas have relatively little olivine and are made up mainly of pyroxene, plagioclase and iron oxide minerals (magnetite, ilmenite). Crystals that have settled from a magma form a cumulate either at the chamber bottom, wall or roof.





Crystal morphology

In rocks crystallised from e.g. a granitic magma, the main minerals are Na-plagioclase, K-feldspar and Quartz, which can crystallise together at -the same time in the plutonic rocks, forming a granular intergrowth in which none of the minerals are euhedral, and most are anhedral (the texture is called a hypidiomorphic granular texture). Some hornblende or mica may be present and as these minerals started to crystallise rather earlier they are often subhedral.

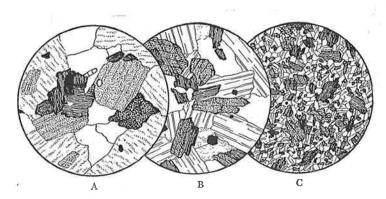
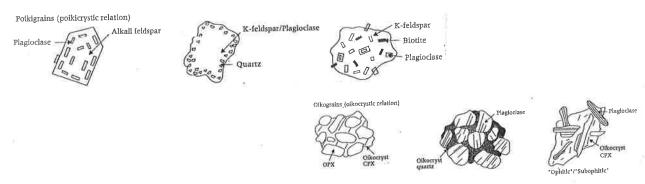


Figure . Granite and Granodiorites

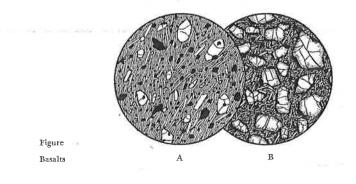
Intergrowth and enclosure

As the minerals crystallising together begin to 'compete' with one another, an intergrowth can occur. If one mineral is growing more quickly it can surround and enclose the other. Partial enclosure results in an **Ophitic texture** - complete enclosure results in a **Poilcilitic texture**. These textures are seen in slowly cooled plutonic rocks such as gabbros. In more rapidly cooled rocks such as in dykes, sills and other shallow intrusions, ophitic texture is the most common - it used to be called the "doleritic" texture, as many basaltic dykes are doleritic.

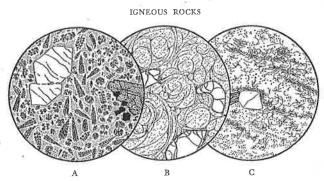


Volcanic

Lavas which erupt to the surface don't show ophitic textures; they cool much more quickly and usually contain a scattering of larger crystals which formed at depth, known as phenocrysts set in a fine-grained groundmass or matrix made up of all the remaining minerals (porphyritic texture). These groundmass crystals have crystallised rapidly from a super-cooled liquid and often form a mesh of plagioclase crystals (long and narrow), in the interstices of which are found rounded granules of pyroxene and iron oxides. This is an interstitial texture.



Erupted rhyolites - may carry phenocrysts of quartz or feldspar, but the liquid is much cooler than the basalt liquid and commonly freezes before fully crystallising. The groundmass may thus contain some very small crystallites, but is often mainly glass (glass is isotropic). The glass is unstable over geological time and will eventually devitrify to give a dense fine-grained mat of the residual minerals - quartz and feldspar in the rhyolites. In the devitrification process, firstly shrinkage produces curved cracks - Perlitic cracks and then crystal growth starts from scattered nuclei at which radiating clusters of tiny crystallites of feldspar form spherulites.



Rhyolitic Pitchstones with Microlites and Crystallites

igneous Petrography Glossary

Crystallinity

holohyaline 100% glass hypocrystalline Holocrystalline 100% crystal

Granularity

Hybabyssal Volcanic Plutonic 1-5mm >5mm Medium Coarse Fine

crystals differ substantially in size e.g. porphyritic approximately the same size inequigranular – equigranular –

unidentifiable in thin section identifiable in thin section cryptocrystalline microcrystalline -

Crystal morphology

regular array of fibres sharing a common optical orientation crystals completely bound by its characteristic faces. round dissolution holes on crystal surface crystals bound by only some of its faces. lack of any characteristic faces. hollows and gaps embayments subhedral – dendritic – euhedral – anhedral – skeletal –

(i.e. 1 crystal) sieve textured crystal –

small, interconnected, box shaped crystal - spongy appearance

Textures

relatively large crystals (phenocrysts) are surrounded by finer porhyritic texture with phenocrysts are bunched/ clustered in grained crystals of the groundmass. glomeroporphyritic texture porphyritic texture –

aggregates/ clots called glomerocrysts. means 1 type of mineral in a clot glomerocryst -

relatively large crystals of one mineral encloses numerous smaller (enclosing crystal) - host crystal crystals (randomly orientated)

polkilitic texture -

enclosed crystal chadacryst

oikocryst -

ophitic texture -

randomly arranged chadacrysts are elongated and are wholly or partly enclosed by oikocryst e.g. plagioclase surrounded by subequant augite.

basis of material occupying angular spaces between feldspar laths. glass or hypocrystalline material partially or wholly occupies

spaces between laths are occupied by 1 or more grains of Px (± OI wedged shaped indices. Glass can be altered.

intergranular -

Intersertial –

interstitial –

sub-parallel arrangement of microcrystalline lath shaped feldspars sub-parallel arrangement of tabular, bladed or prismatic crystals in groundmass of holo or hyalocrystalline rock and opaques)

plagioclase in feldspar or vice versa visible to naked eye. perthitic/antiperthitic -

trachytoid texture – trachytic texture -

corona texture -

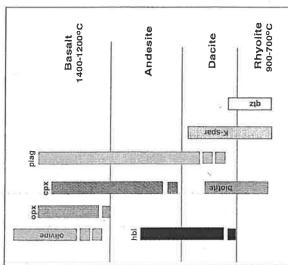
crystals of another mineral e.g. ol surrounded by opx or hbl by biot. crystals of one mineral surrounded by rim/mantle of 1 or more

Zoning

solid solution composition from rim to core high to low e.g. An → Ab rich plagioclase low to high e.g. Ab → An plagioclase gradual changes abrupt changes discontinuous continuous reversed – zoning – normal -

Mineralogy

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Hibbard M.J. Petrography to Petrogenesis. Prentice Hall MacKenzie W.S., Donaldson C.H., Guilford C. Atlas of Scientific and Technical
MacKenzie W.S, Guilford C. Atlas of rock forming
minerals in thin sections. Longman Scientific and Igneous rocks and their textures. Longman Gribble C.D., Hall A.J. A Practical Introduction Technical.

- SUMMARY TABLE -

Thin section Slide No.	OLIVINE	NHO NH	CPX	AMPHIBOLE	BIOTITE	MUSCOVITE	PLAGIOCLASE	ALK. FELDS.	QUARTZ	ORE	ROCK NAME + CONMENTS
	4						× ,				-
								7.1			
	-										
		· ·					•				83
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200											Diameter Control of the Control of t
									13		

Homework 1 Geochemistry / Igneous Petrology

Petrogenetic modelling using mass balance calculations

$$C_j = \sum_{i=1}^n \alpha_i M_{i,j}$$
 equation 1

in major oxide terms this equation translates as: $C_j = \text{concentration (\%) of element j in the rock,}$ $M_{i,j} = \text{concentration (\%) of element j in component i (a component)}$

could be a mineral, a rock composition, an end-member magma),

 $\alpha_i = \text{fraction of component i}$ n = number of components

PROCESS	COMPOSITION	NOI	FRACTION
	び	Mil	α_{i}
Magma mixing	Hybrid rock	End-member 1 End-member 2	α (1 - α)
Fractional	Parental magma	Extracted crystals	Σαį
crystallisation		Residual magma	(1 - 2ai)

Exercise (4) Magma mixing

You are given the composition of a dacite magma, and your hypothesis is that it is the result of mixing between an andesite and a rhyolite magma. Try to find the proportions (α in the equation above) of the two end-member magmas involved in magma mixing. Examine the table (overleaf), and estimate (?guesstimate) the likely proportions of each end-member magmas that you would require to mix together to approximate the composition of the mixed magma (dacite).

Rearrange equation (1) to express your magma mixing model and calculate the results for each oxide. Estimate the "goodness of fit" using r^2 as indicated (a value of < 1.0 is usually taken as acceptable).

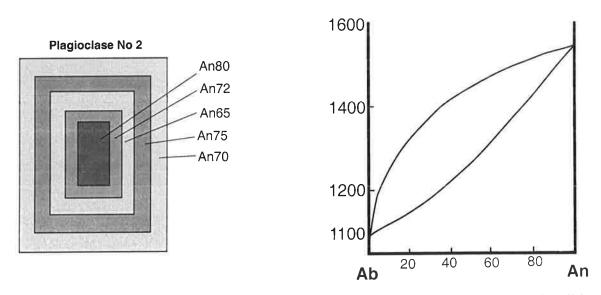
Note that DACITE_{obs} is the analysed composition of the dacite while DACITE_{calc} is the estimated composition derived by blending appropriate proportions of rhyolite and andesite. Find the values of α for andesite and rhyolite that gives your best match for DACITE_{obs}.

r ² (OBS - CALC) ²											= \rac{1}{2}r^2 =
calculated mix DACITE _{calc}									30		Goodness of fit = Σr^2 =
mixed magma o	65.98	0.59	16.15	2.47	2.33	1.81	4.38	3.85	2.20	0.15	
end-member 2 ANDESITE	58.70	0.88	17.24	331	4.09	3.37	6.88	3.53	1.64	0.21	
end-menber l RHYOLITE	73.95	0.28	13.48	1.50	1.13	0.40	1.16	3.61	4.37	. 0.07	
4	SiO,	TiO,	AJ,Ó,	Fe O	, O	MPO	, Ç	Na,O	K,O	P.O.]

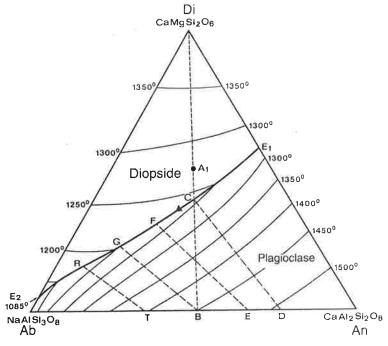
Exercise (2) Fractional crystallisation

If 30% of a diorite magma crystallises as amphibole and 25% as plagioclase then find the composition of the residual melt for each oxide.

пем таета		ı,			<u>.</u>					
Plagioclase	58.10	0.00	26.44	0.04	0.15	0.03	7.84	6.48	1.10	0.00
Amphibole	44.99	1.46	11.21	3.33	13.17	10.41	12.11	0.97	0.76	0.00
Diorite	57.48	0.95	16.67	2.50	4.92	3.71	6.58	3.54	1.76	0.29
X 0	SiO,	Tio,	ĄŢĠ	Fe.O.		MEO	, O	Na,O	Ж O	P,0.



1. Describe the crystallisation history of plagioclase crystal No 2, using the An-Ab solid solution phase diagram. What has happened in the magma chamber? Write a brief outline of your explanation. Note there is more than one possible scenario!



2. Consider equilibrium crystallisation for a starting composition of 30% $An_{40}Ab_{60}$ and 70% Di in the system Di-An-Ab. Estimate the composition of the final liquid?

Homework 3 Geochemistry / Igneous Petrology

1. Partition coefficients and bulk distribution coefficients

The distribution of a trace element between silicate liquid and crystal is described by the partition (or distribution) coefficient:

$$K_{D_{j}} = \frac{C_{j, mineral}}{C_{j, melt}}$$

For an assemblage of minerals crystallising from a magma the bulk distribution coefficient. (D) is

$$D_{j} = \sum_{i=1}^{n} w_{i} K_{D_{i,j}}$$

where w; is the mass fraction of mineral i,

and KDi, is the partition coefficient for element j in mineral i.

Taking the example of Ce in a peridotite comprising of 60% olivine, 30% opx, 5% cpx and 5% garnet. The KD values for these minerals in basic melt are given in the table below:

	w		KD _{Ce}	$K_{D_{Ni}}$				
Olivine	0.60		0.007	10				
Opx	0.30		0.02	5				
Срх	0.05		0.15	8				
Garnet			0.03	0.01				
	1.00							
D _{Ce} =	0.60 x 0.007	+	0.30 x 0.02	+	0.05 x 0.15	+	0.05 x 0.03	
Ce	olivine		opx		cpx		garnet	
THE STATE OF THE S	0.019							

Exercise

Calculate the bulk distribution coefficient for Ni (D_{Ni}) in the same peridotite.

2. Spidergrams

The so-called spidergrams are line plots of the ratio:

rock concentration reference composition

of a range of elements plotted in a particular sequence along the X-axis (see Rollinson section 4.4 or Wilson p19 for explanation). The ratio is plotted on the Y-axis with a logarithmic scale (0.1, 1, 10, 100, 1000 etc., as appropriate).

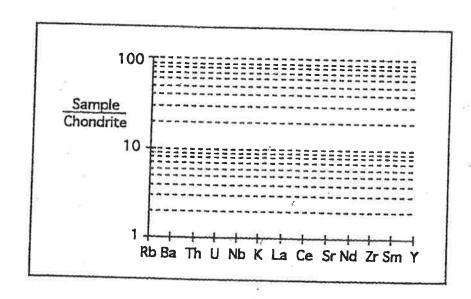
The normalisation or reference standard may be a chondrite (to represent bulk earth composition) or MORB, or other source rock compositions. Rollinson (p143) has a list of appropriate reference values as well as lists of various X-axis element combinations.

Exercise

Compare two types of basalt from contrasting settings by plotting their chondrite-normalised spidergrams on the same graph. The basalts are a mid ocean ridge basalt of "normal" composition (N-type MORB) and a tholeiitic basalt from an island arc (IAT).

Plot your ratios on the template provided below. (Abundances for trace elements are normally quoted in parts per million, or ppm).

Element	N-type MORB	Island arc basalt	Chondrite	
Rb	1.0	14	0.35	
Ba	12	300	6.9	
<u>Ba</u> Th	0.20	1.1	0.042	
U	0.10	0.36	0.013	
Nb	3.1	1.4	0.35	
K	1060	8640	120	
La Ce	3.0	10	0.328	
Ce	9.0	23	0.865	
Sr	124	550	11.8	
Vd	7.7	13	0.63	
<u>r</u>	85	40	6.84	
m	2.8	2.9	0.20	
	29	15	2.0	



Briefly describe the differences between the two samples and outline one or more possible reasons for these differences.

The behaviour of trace elements during partial melting can be modelled in two end-member cases. (In nature the equilibrium case is probably closer to reality.)

Batch melting model

Melting is regarded as taking place as a batch in equilibrium with residue before removal of any melt.

$$\frac{C_L}{C_0} = \frac{1}{D(1-F)+F}$$

where $C_0 = initial$ composition of source rock,

C_L = concentration of element in liquid (melt),

D = bulk distribution coefficient for that element,

and F = fraction of melt (i.e. 1= all melt, 0 = all solid as crystal residue, no melt).

Perfect fractional (Rayleigh) melting model

In this case each infinitesimally small increment of melt is removed from contact with the source rock. This is physically unrealistic but represents an end-member case.

$$\frac{C_{L}}{C_{0}} = \frac{1}{D} (1 - F)^{(\frac{1}{D} - 1)}$$

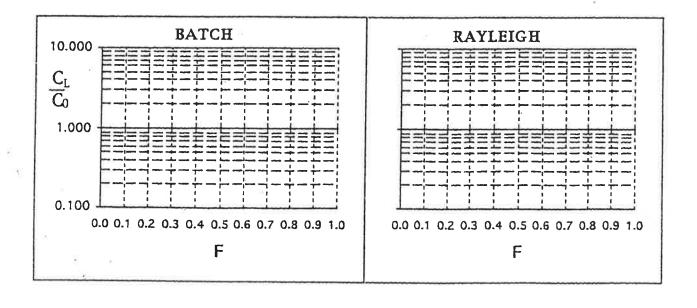
Exercise

Calculate the C_L/C_0 ratios for both the batch and Rayleigh melting of an incompatible element with D = 0.1 and a compatible element with D=10.

Perform the calculations for 10%, 30%, 50%, 70%, and 90% melting and complete the table below.

	BATO	CH	RAYLEIGH		
F	$\frac{C_L}{C_0} (D=0.1)$	$\frac{C_{L}}{C_{0}} (D=10)$	$\frac{C_L}{C_0} (D=0.1)$	$\frac{C_L}{C_0} (D=10)$	
0.1					
0.3					
0.5					
0.7					
0.9					

Graph the results on the templates (one each for equilibrium and Rayleigh):



RADIOGENIC ISOTOPES

A radiogenic isotope system is one where a daughter isotope produced through radioactive decay of a parent isotope. Normally the daughter isotope is considered in relation to a stable isotope of the same element.

The basic radioactive decay equation may be expressed as:

$$D = D_0 + N(e^{\lambda t} - 1)$$
 (equation 1)

where

D = total no. of daughter atoms,

 $D_0 = \text{no.of radiogenic daughter atoms present initially,}$

N = no. of parent atoms remaining,

t = time,

$$\lambda = \text{decay constant } (\lambda = \frac{\ln(2)}{T_{\frac{1}{2}}} = \frac{0.693}{T_{\frac{1}{2}}} \text{ where } T_{\frac{1}{2}} = \text{half life})$$

Equation 1 may be arranged as:

$$.87Sr = .87Sr_i + .87Rb(e^{\lambda t} - 1)$$
 (equation 2)

where ⁸⁷Sr and ⁸⁷Rb are the present day values and ⁸⁷Sr_i is the initial value before the system closed and decay began.

As 86Sr is a stable isotope this will not change with time, and we may divide through each term in equation 2 by this factor.

$$\frac{{}^{87}\!Sr}{{}^{86}\!Sr} = \left(\frac{{}^{87}\!Sr}{{}^{86}\!Sr}\right) + \frac{{}^{87}\!Rb}{{}^{86}\!Sr} \left(e^{\lambda t} - 1\right) \dots (equation 3)$$

This equation is the basis for Rb-Sr geochronology.

When the age (t) is accurately known, then the initial ratio for the Sr isotopes (variously written as $\left(\frac{87}{86}\text{Sr}\right)_{i}$, $\left(\frac{87}{86}\text{Sr}\right)_{0}$, or $\left(\frac{87}{86}\text{Sr}\right)_{t}$ where t is the age in Ma) can be more accurately calculated for each sample using the following rearrangement of equation 3:

$$\left(\frac{8^{7}Sr}{8^{6}Sr}\right) = \frac{8^{7}Sr}{8^{6}Sr} - \frac{8^{7}Rb}{8^{6}Sr} \left(e^{\lambda t} - 1\right)$$
(equation 4)

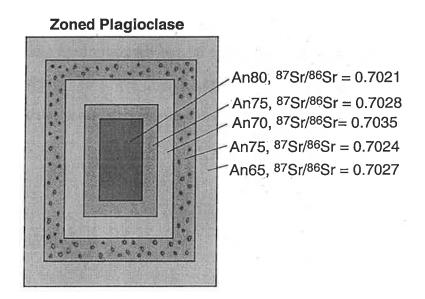
Exercise 1:

Assuming an age (t) of 430Ma and using the decay constant for the Rb-Sr system ($\lambda = 1.42 \times 10^{-11} \text{ y}^{-1}$, find the initial ratios for the following 4 granite samples (equation 4):

Sample	87Rb/86Sr	⁸⁷ Sr/ ⁸⁶ Sr	$\left(\frac{87}{86}Sr\right)$
A B	0.1973	0.70607	S 10
C	1. 73 8 1. 68 6	0. 72 080 0. 72 049	
D	0.9055	0.71373	

Exercise 2:

The plagioclase crystal below was analysed for major elements (An mol%) and for Sr isotope composition of its individual zones. Plot its An mol% and its Sr isotope composition vs. the position in the crystal (e.g. core, zone 1, zone 2 ... on x-axis). Briefly discuss possible causes for the variations observed.



Homework No 6 Geochemistry/Igneous Petrology

Contrasting radiogenic isotopes, that do not change by physical influence (Δ P, T) or fractional crystallisation, oxygen isotopes fractionate during partial melting and during crystalliquid fractionation. This is because the mass difference between oxygen isotopes is much bigger due to their smaller weight, relative to the mass difference of e.g. Nd isotopes. However, at magmatic temperatures this fractionation is small (ca. 0.2‰ per 5wt% SiO₂).

Therefore primitive basaltic magmas should have an isotope composition reflecting their mantle source (5.5-6‰). Evolved rocks such as granites and rhyolites should hence be enriched in δ^{18} O by ca. 1‰ assuming closed system differentiation and a SiO₂ increase of ca. 25 wt%.

To assess differences between end member compositions within a suite of samples the difference (Δ) of their oxygen isotope values is commonly used. For example the difference between a rhyolite and a basalt can be expressed as $\Delta_{\text{rhyolite-basalt}} = \delta^{18} O_{\text{rhyolite}}$. This method is also used to evaluate differences between mineral pairs that help constrain oxygen fractionation throughout a sample suite.

Exercise 1.

The following five samples from a recently erupted volcano yield δ^{18} O values that are markedly different. Calculate the $\Delta_{\text{rhyolite-basalt}}$ and briefly discuss the likelihood of the suite having evolved in an open vs. a closed system.

Sample	SiO ₂	$\delta^{18}O_{wr}$	$\delta^{18}\mathrm{O}_\mathrm{px}$	$\delta^{18} O_{fsp}$	$\Delta_{ ext{fsp-px}}$
4					
Basalt	46	5.6	5.3	5.5	
Basaltic Andesite	53	6.1	5.8	6.0	
Andesite	59	6.6	6.2	6.5	
Dacite =	65	6.9	6.6	6.9	
Rhyolite	72	7.4	7.0	7.3	

Exercise 2.

Calculate the Δ_{fsp-px} values of the five samples and comment on the results (i.e. are they consistent with the results you derived from the whole rock data?).

